Preliminary Studies of the Obtaining of Solid Metallic Cerium from Fluoride Melts

Virgil Constantin, Ana-Maria Popescu, and Stefania Zuca

Romanian Academy, Institute of Physical Chemistry "I. G. Murgulescu", Splaiul Independentei 202, Bucharest 77208 - Romania

Reprint requests to V. C.; Fax: +4-01-3121147; E-mail: virgilconstantin@yahoo.com

Z. Naturforsch. **58a**, 57–62 (2003); received July 28, 2001 / September 11, 2002

Presented at the NATO Advanced Study Institute, Kas, Turkey, May 4-14, 2001

The present study deals with the obtaining of solid cerium by molten salt electrolysis of a 46.74 - 48.26 - 5wt% LiF-NaF-NaCeF₄ mixture, in the temperature range 700 - 730°C and with a current efficiency of ~ 75 %.

For this purpose NaCeF₄ was obtained, characterized and it's cubic form was identified. The solubility and decomposition potential of this compound in the molten electrolyte 49.2 - 50.8% LiF-NaF was also studied.

Key words: Cerium; Electrolysis; NaCeF₄ Decomposition Potential; Molten Salts.